

A = reactants B = products

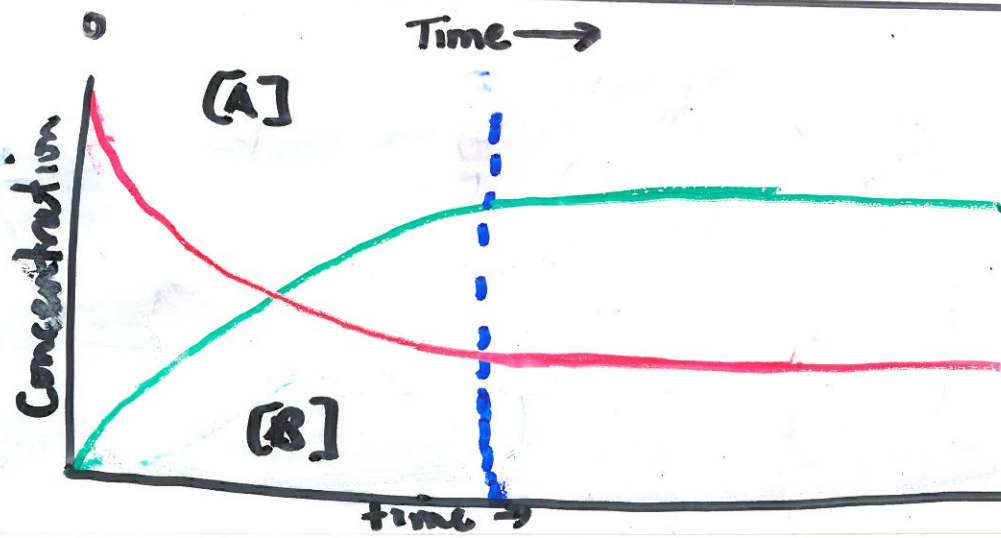
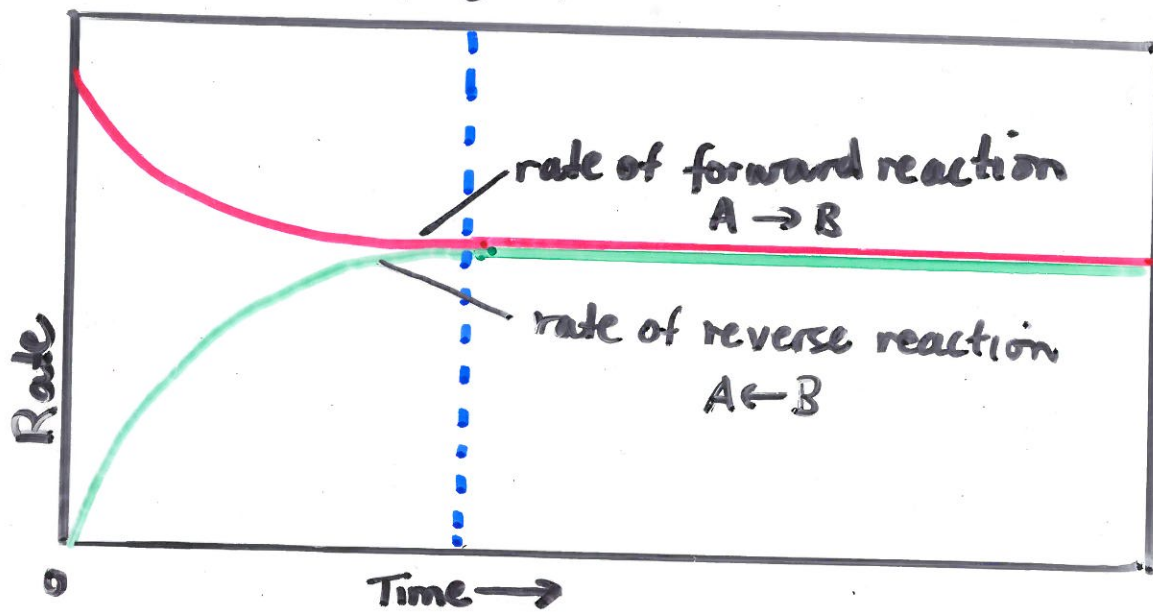
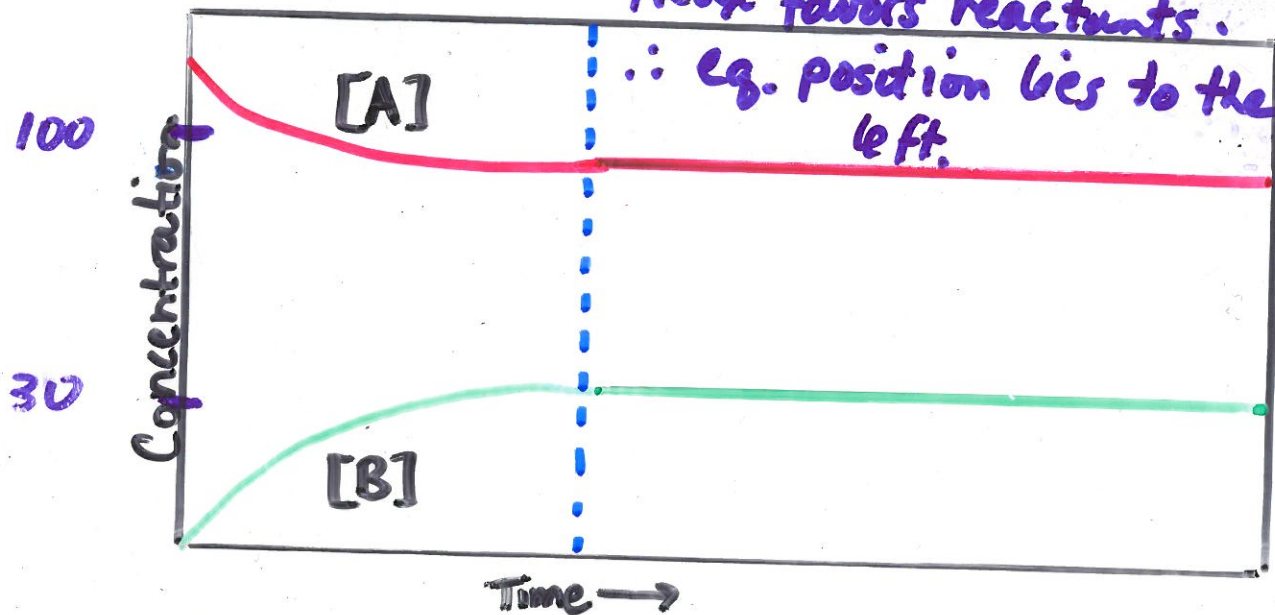
Changes As a Reaction Approaches



Equilibrium

Reax favors reactants.

\therefore eq. position lies to the left.



* This reaction favors products.
 \therefore equilibrium position lies to the right.